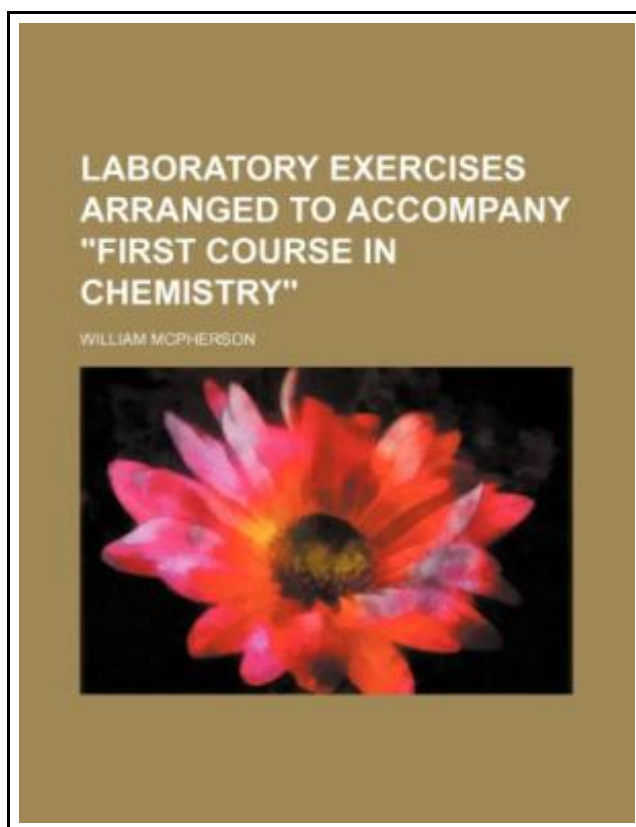


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Rarebookclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1915 Excerpt: .poisonous gas. All of the experiments must be performed in the hood. After the gas is generated, pour the contents of the generator flask into a sink or jar in the hood. (Two students may work together.) Arrange an apparatus according to Fig. 38. B contains a solution of sodium hydroxide, and C and D solutions of limewater. The tube E contains near its middle a layer of 1 or 2 g. of copper oxide. The whole apparatus must be air-tight. After the apparatus has been approved by the assistant, put 7 or 8 g. of oxalic acid ($C_2H_2O_4$) into the flask A and pour over this 25 cc. of sulfuric acid. Stopper the flask tightly and apply a very gentle heat, at the same time heating the copper oxide in E. Regulate the heat so as to cause the evolved gas to bubble slowly through B and C. Collect over water any gas escaping from D. What evidences have you that the oxide is reduced? Finally, withdraw the heat from A and at once remove the stoppers from A, B, and C, and test with a burning splint the gas inclosed in each. Account for the results. Also test with a burning splint the gas which escaped from D and account for the result. Why is the gas passed through a solution of sodium hydroxide? Write the equations for the reactions which take place in A, B, C, D, E. EXERCISE 50 A STUDY OF...



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